# **Crystal System**



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#### Solid State

Crystalline

Atoms Stacked
in periodic
manner
Repeated
pattern unit in 3
dimensions

#### Polycrystalline

 Periodicity is interrupted at grain boundaries

#### Amorphous

•Size of grains comparable to pattern unit

# **Crystal Structure**

An ideal crystal is formed by infinite repetition of an identical group of atoms.

>Three dimensional periodic array of identical building blocks

➢ Basis:-Atoms, group of atoms

>Lattice:- Set of mathematical points to which basis is attached





### Crystal structure

# **Bravais and Non Bravais Lattice**



**Bravais Lattice** 

All lattice points are equivalentAll atoms are of the same kind



#### Non Bravais Lattice

- •Some of the lattice points are nonequivalent
- •Translation from A to A' is not equivalent
- •Lattice with a basis or interpenetrating bravais lattices

# **Lattice Translation Vectors**

- Lattice is invariant under translation
- i.e. lattice as viewed from any lattice point remains invariant, translated by integral multiple of basis vectors **a,b,c**
- **R'**= **R**+  $n_1$ **a**+ $n_2$ **b**+ $n_3$ **c** = **R**+**T**
- Where n<sub>1</sub>,n<sub>2</sub>,n<sub>3</sub> are arbitrary integers, **T** is known as translation vector



## **Basis Vectors and Primitive Unit Cells**

•Smallest unit which when translated by all lattice vectors covers the lattice once and only once.

Origin

- •The vectors a and b are basis vectors
- •Choice of unit cell is not unique
- •Each unit cell has only one lattice point
- All unit cells have same area
  S=|a×b|



## Wigner Seitz Unit Cell

- Weigner Seitz unit cell about a lattice point is the region of space that is closer to that particular point than to any other lattice point.
- •Contains only 1 lattice point per unit cell
- •Has all the properties of Primitive Unit cell.

### Construction

- $\checkmark$  Join all lattice points by straight lines
- ✓ Draw normal bisector line/plane to these lines
- ✓ The smallest volume enclosed in this way is the Wigner Seitz primitive cell.
- ✓When translated by all translation vectors complete lattice can be covered by this cell only once.

## Non Primitive Unit Cell

•Unit cell with area larger than primitive unit cell•More than 1 lattice point per unit cell

Non-primitive cells and non-Bravais lattices have no connection. Non Primitive Cell refers to the particular (and somewhat arbitrary) choice of basis vectors in a Bravais lattice. Non Bravais Lattice refers to the physical fact of nonequivalent sites.





## Lattice types in 2 dimensions

Lattice Type	Conventional unit cell	Axes and angles	Point group symmetry about lattice point
Oblique	Parallelogram	a≠b, ø=0°	2
Square	Square	a=b, ø=90°	4 mm
Hexagonal	60° rhombus	a≠b, ø=120°	6 mm
Primitive rectangular	Rectangle	a≠b, ø=90°	2 mm
Centred rectangular	Rectangle	a≠b, ø=90°	2 mm

## Lattices in 2 D







(d) Centred rectangular lattice axes are shown for both the primitive cell and the rectangular unit cell for which  $|\mathbf{a}| \neq |\mathbf{b}|, \gamma = 90^{\circ}$ 

### 7 crystal Systems, 14 Bravais Lattices in 3D

System	Essential Symmetry	Bravais Lattices	Restrictions on conventional cell angles and axes
Triclinic	No Planes, No Axes	Simple	a≠b≠c α≠β≠Υ
Monoclinic	One 2-fold axis or one plane	Simple Base centred	a≠b≠c α=Ƴ=90°≠β
Orthorhombic	Three mutually perpendicular 2-fold axis or 2 planes intersecting in a 2-fold axis	Simple Body centred Base centred Face centred	a≠b≠c α=β=۲=90°
Tetragonal	One 4-foldaxis or 4-fold inversion axis	Simple Base centred	a=b≠c α=β=Ƴ=90°
Cubic	Four 3-fold axis	Simple Body centred Face centred	a=b=c α=β=۲=90°
Trigonal	One 3-fold axis	Simple	a=b=c α=β=Ƴ<120°,≠90°
Hexagonal	One 6-fold axis	Simple	a=b≠c α=β=90° Ƴ=120°



Triclinic

Monoclinic

Monoclinic base centred









Ortho-rhombic

Ortho-rhombic base centred

Ortho-rhombic body centred

Ortho-rhombic face centred



Cubic

## References

- Introduction to Solid State Physics, Charles Kittel
- Elementary Solid State Physics, Principles and Applications, M. A. Omar
- Solid State Physics, A. J. Dekker
- Solid State Physics, M.A. Wahab
- Elements of Solid State Physics, J. P. Srivastava



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